



**PARVATHANENI BRAHMAYYA  
SIDDHARTHA COLLEGE OF ARTS & SCIENCE**  
*Autonomous*  
Siddhartha Nagar, Vijayawada-520010  
*Re-accredited at 'A+' by the NAAC*

### **CONDENSEDMATTERPHYSICS-I(Special)**

Offered to : M.Sc.(PHYSICS)	Course Code : 22PH3D3
Course Type : Domain specific elective (DSE)	Course : CONDENSEDMATTERPHYSICS-I(Special)
Year of Introduction : 2004	Year of offering : 2022
Year of Revision : 2022	Percentage of Revision : Nil
Semester : III	Credits : 4
Hours Taught: 60 hrs. per Semester	Max.Time : 3 Hours

**Course Description:** Condensed Matter Physics -I is the field of physics that deals with the macroscopic and microscopic physical properties of matter, especially the solid and liquid phases which arise from electromagnetic forces between atoms. More generally, the subject deals with "condensed" phases of matter: systems of many constituents with strong interactions between them.

#### **CourseObjectives:**

1. To understand the lattice defects in different crystals
2. To understand the thermal and optical properties of crystals due to lattice defects.
3. To learn the concepts of luminescence and phosphorescence
4. To understand the specific heat of solids in different crystals
5. To understand the magnetic properties and different theories of magnetism

**CourseOutcomes:** At the end of this course, students should be able to:

CO1: Remember the concepts of crystal structures and their properties

CO2: Understand the importance of crystal defects

CO3: Analyze the process involved in the Luminescence.

CO4: Understand the importance of specific heat of solids.

CO5: Analyze the theories involved in different magnetic domains.

<b>CO-PO MATRIX</b>								
	<b>CO-PO</b>	<b>PO1</b>	<b>PO2</b>	<b>PO3</b>	<b>PO4</b>	<b>PO5</b>	<b>PO6</b>	<b>PO7</b>
22PH3D3	<b>CO1</b>	H					L	M
	<b>CO2</b>	H					L	M
	<b>CO3</b>			H			L	M
	<b>CO4</b>	H					L	M
	<b>CO5</b>			H			L	M

<b>Syllabus</b>		
<b>Unit</b>	<b>Learning Units</b>	<b>Lecture Hours</b>
I	<p><b>Crystal Defects</b> The structure of metals, lattice defects and configurational entropy – The number of vacancies and interstitials as function of temperature, the formation of lattice defects in metals, interstitial diffusion in metals, chemical diffusion in metals-Kirkendall effect, Edge and screw dislocation, Estimates of dislocations densities, The Frank - Read mechanism of dislocation multiplication. (CO1)</p>	12
II	<p><b>Optical Properties</b> Optical and thermal electronic excitation in ionic crystals. The ultraviolet spectrum of the alkali halides; excitons, Illustration of electron-hole interaction in single ions, Qualitative discussion of the influence of lattice defects on the electronic levels, Nonstoichiometric crystals containing excess metal. The transformation of F centers into F<sup>+</sup>-centers and vice-versa, Photoconductivity in crystals containing excess metal, Color centers resulting from excess halogen, Color centers produced by irradiation with X-rays. (CO2)</p>	12
III	<p><b>Luminescence</b> Introduction, Kinds of Luminescence, Excitation and emission, Efficiency of Phosphor, Decay mechanisms, Thermo luminescence and glow, Thallium-activated alkali halides, the sulfide phosphors, Electroluminescence. (CO3)</p>	12
IV	<p><b>Lattice Vibrations and Thermal Properties</b> Elastic waves in one dimensional array of identical atoms. Vibrational modes of a diatomic linear lattice and dispersion relations. Acoustic and optical modes. Infrared absorption in ionic crystals. Phonons and verification of dispersion relation in crystal lattices. Lattice heat capacity – Einstein and Debye theories. Lattice thermal conductivity - Phonon mean free path. Origin of thermal expansion and Grüneisen relation. (CO4)</p>	12
V	<p><b>Magnetic Properties of Solids</b> Quantum theory of Paramagnetism, Crystal Field Splitting, Quenching of the orbital Angular Momentum Ferromagnetism Curie point and the Exchange integral, Saturation Magnetization at Absolute Zero, Magnons, Bloch's <math>T^{3/2}</math> law. Ferromagnetic Domains. Ferrimagnetism. The structure of ferrites, The saturation magnetization, Elements of Neel's theory. (CO5)</p>	12

### Reference Books:

1. A.J. DEKKER, Solid State Physics, Macmillan, 2002
2. CHARLES KITTEL, Introduction to Solid State Physics, John Wiley & Sons, 2007
3. GUPTA and KUMAR, Solid State Physics, K. Nath & Co., 2000
4. S.O. PILLAI, Solid State Physics New Age International, 2006
5. M.A. Wahab, Solid State Physics, Narosa, 2019, 3<sup>rd</sup> edition